Oxyfuel Conversion of Heaters and Boilers for CO$_2$ Capture

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Abstract

The CO$_2$ Capture Project is a major joint energy initiative by the energy industry to address problems of CO$_2$ emission that have the potential to cause climate change. Its objective is to provide very significant cost reduction in CO$_2$ removal and storage compared to existing technologies by considering various technology options applied at different real locations.

This feasibility study involves the potential application of Oxyfuel technology on a refinery-wide basis at the BP Grangemouth unit in Scotland. Previous studies in Oxyfuel technology have considered application for retrofit on a power station boiler and refinery process heaters. A total of seven boilers and 13 process heaters of various types burning a mixture of refinery fuel gas and fuel oil form the basis of this study.

This work considers the issues involved in locating two world scale air separation plants totalling up to 7400 tonne/day of oxygen plus a CO$_2$ compression and purification system on a congested site, in addition to the issues of distributing the oxygen around the site and collecting the CO$_2$ rich effluent from the combustion processes for purification and final compression.

1.   Introduction

1.1 The CCP

The CO$_2$ capture project (CCP) is a major joint energy industry effort to respond to concerns which exist around climate change and CO$_2$ concentrations in the atmosphere. The CCP involves the following participant companies: BP (Co-ordination), Chevron Texaco, En Cana, Eni, Norsk Hydro, Shell, Statoil and Suncor.

Funding comes from, not only the participating companies, but also from the Department of Energy in the U.S., from Norway and from the European Union.

The function of the CCP is to achieve major reductions in the cost of CO$_2$ capture and storage compared to existing technologies. In addition, there is work to show that geological CO$_2$ storage is safe, measurable and verifiable.

The mechanism of the project has been to commission studies on various technology options applied to real locations where major CO_2 emissions occur. The objective is to reach proof of concept stage by the end of 2003 and have a demonstration in operation by 2010.

1.2 Technology Options

There are 3 main technology options for the removal of CO_2 from fossil fuel combustion processes.

1. The end of pipe solution involving using an absorption or adsorption process for CO_2 removal from the vent gas. The base case for these studies is the use of an amine chemical absorption system.
2. Precombustion decarbonisation involves the conversion of the fossil fuel to hydrogen with all carbon converted to CO_2, which can be removed from the pressurised hydrogen fuel gas stream prior to its use as fuel for power production in a gas turbine combined cycle or a fuel cell.
3. Oxyfuel, which involves the substitution of air in the fossil fuel burner with pure oxygen and recycled flue gas, giving a stream which, after cooling and condensation of water, consists of a high concentration of carbon dioxide.

The produced CO_2 following compression and delivery by pipeline can be disposed of in geological formations as follows:

1. Injection into oil formation for enhanced recovery.
2. Storage in depleted oil and gas reservoirs.
3. Storage in saline aquifers

1.3 The Study of The Oxyfuel Option

This paper presents preliminary results on a feasibility study at present being carried out on the oxyfuel conversion of steam raising boilers and process heaters in the Grangemouth refinery of BP which is located in Scotland near Edinburgh. The study is being carried out by Air Products as technology provider to the CCP with the assistance of Mitsui-Babcock (boiler conversions) and Foster Wheeler (process heater conversions).

2. CO_2 Emissions from the Grangemouth Refinery

A number of studies have been published (1,2,3) indicating that conversion of existing steam boilers and process heaters to oxyfuel firing is feasible at low cost and often with improved performance. Projected overall costs which include oxygen supply and CO_2 processing and compression are competitive with other CO_2 capture technologies. Oxyfuel is pre-eminently suited to retrofit conversion of existing fossil fuel fired facilities. The fuel in this case is refinery fuel gas, plus refinery fuel oil.

The task for this study is to consider the total equipment system including associated services for extra power, cooling water, etc., to convert a number of units in the BP Grangemouth refinery to oxyfuel firing. The CO_2 produced at each unit will be collected, purified, compressed and delivered at 220 barg pressure and at a suitable purity for pipeline transportation. The objective is to produce 2 million tonnes of CO_2 per year.

Oxygen supply for this study will be from two cryogenic air separation plants. A future task will be to consider the use of high temperature ion transport membranes for oxygen production.

The list of units which will be converted for oxyfuel firing is as follows:

Five Simon Carves boilers each supplying 300,000 lb/hr steam. Typical fuel mix 40% gas, 60% oil by weight. These are linked to two stacks.

Two Babcock steam boilers each supplying 500,000 lb/hr steam. Average fuel mix 40% gas 60% oil by weight. These two boilers are linked to a single stack.

The steam conditions for all these boilers is 127.6 barg, 518°C.

There are, in addition, a total of 12 process heaters of various types which have been specified for oxyfuel conversion – box, cabin or vertical cylindrical. Duties vary from 10.3MW to 112.3MW. Fuel is either gas alone or a combination of gas and fuel oil. In addition, there is a hydrogen producing steam/natural gas reformer furnace fired by fuel gas. Summary details of the heaters and boilers considered in this study are given in Table 1. This gives the CO_2 emissions with air firing, the CO_2 delivered to the pipeline when operating in the oxyfuel mode and the total oxygen consumptions. It is clear from these results that one of the benefits of oxyfuel firing is a reduction in fuel required, in this case 6% reduction. This is the reason that the total CO_2 captured is below the 2.0 million tonne per year target.

	Air Firing		Oxyfuel Firing		
	Total Fuel Consumption kg/hr	Total CO_2 Emitted kg/hr	Total Fuel Consumption kg/hr	Total O_2 Consumption kg/hr	Total CO_2 Captured kg/hr
Boilers B1 - B7	54,809	164,270	52,519	179,835	146,424
Heaters H1 - H12	26,510	73,828	24,302	90,339	63,166
Reformer H13	3,600	9,791	2,813	10,494	7,135
		Totals	6736 tonne/day O_2		
			1.90 million tonne/year CO_2		

Table 1: Grangemouth Heaters and Boilers

3. Oxyfuel Boiler Conversion

A detailed analysis of the conversion of one of the Babcock boilers at BP Grangemouth has been given in a previous paper (2,3). Figure 1 shows diagrammatically the way in which the boiler conversion is carried out and the typical performance characteristics of the oxyfuel system.
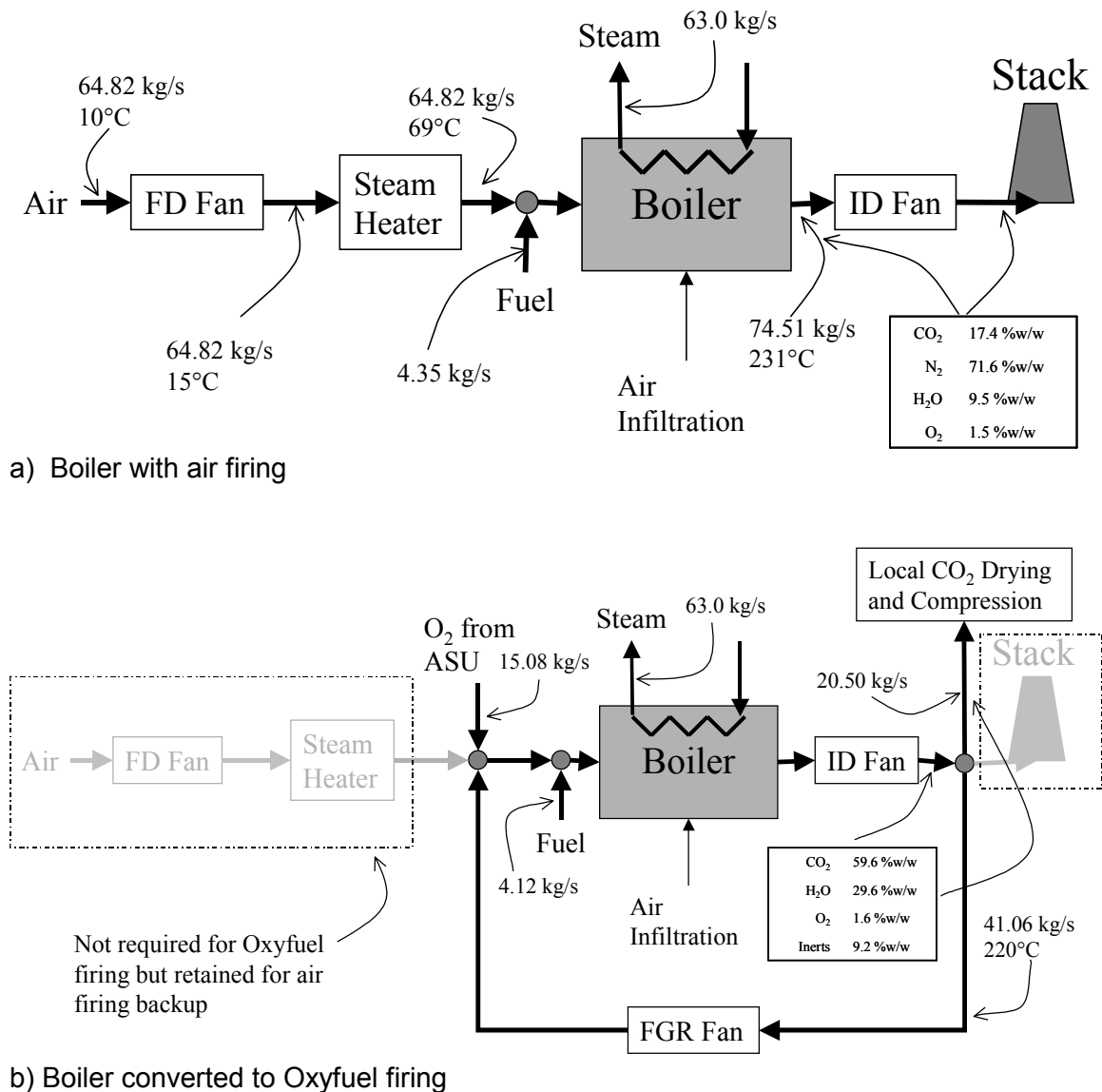

a) Boiler with air firing

b) Boiler converted to Oxyfuel firing

Figure 1: Comparison of air and oxyfuel firing boiler

New equipment will include a recycle flue gas line and blower, 100% shutoff stack damper, oxygen injection and mixing system and possible burner modifications. The revised control system will allow air firing to be re-established in the event of an oxygen supply failure without tripping the boiler. This can be achieved with a liquid oxygen (LOX) and gaseous oxygen (GOX) instant demand back-up system which maintains oxygen supply pressure while the air fans are started.

A critical parameter, which must be established by careful performance analysis on an existing boiler is the amount of air in-leakage and the possibility of reducing this to a minimum by repairs. The inert nitrogen and argon can be a problem, particularly if the CO_2 is to be used for enhanced oil recovery. The limit for these studies is 3%. Excess O_2 plus SO_2, NO_X, CO or hydrocarbons are not considered to be a problem for this application. If the boiler air in-leakage is low, then an oxygen purity of 99.5% would be specified to keep (N_2+Ar) below 3%. If the air in-leakage is higher then 95% to 97% oxygen purity would be specified and an inerts removal plant would be used to reach the required

specification. In all cases considered at BP, tests and engineering analysis have shown that the air in-leakage achievable means an oxygen purity specification of 95% and the provision of an inerts removal plant.
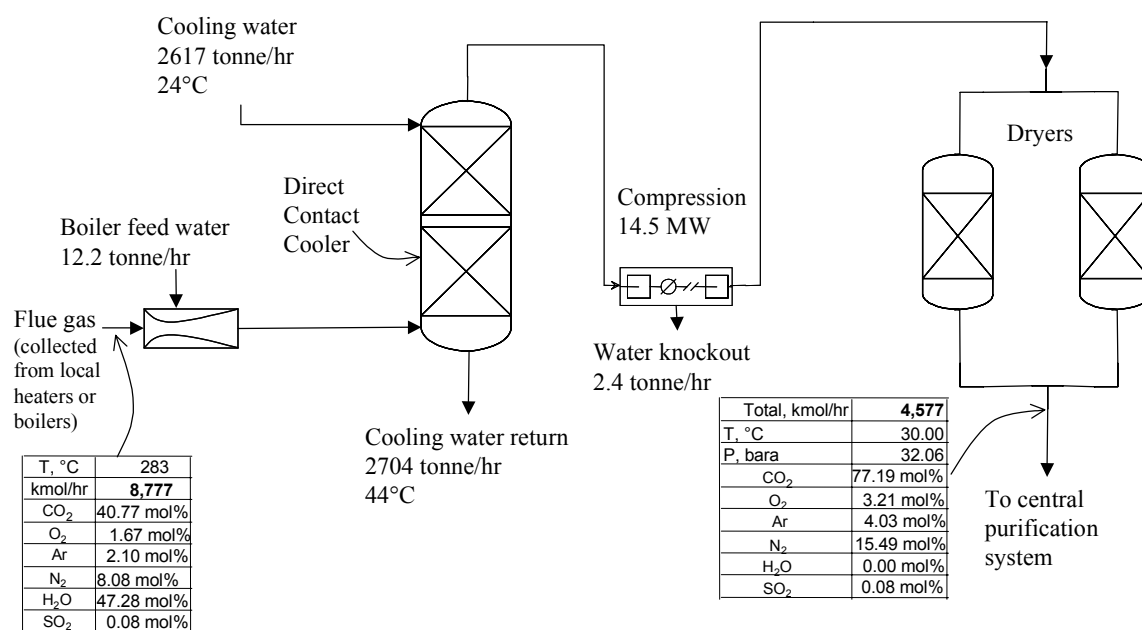
4.   Oxyfuel Heater Conversion

The conversion of process heaters to oxyfuel firing requires a similar modification to the system as that described for the boiler conversion (1). A key criteria for the process heaters is to ensure that the peak heat flux to the tube surfaces is not increased. This is normally fixed by stability of process fluids or tube metallurgy. This constraint is maintained for the higher emissivity CO_2 rich gas in the furnace by increasing the recycle to oxygen feed ratio to operate at below 21% oxygen concentration thus limiting flame temperature. The same overall duty is maintained in each case and also the same balance of radiant and convection section duties. The firing rate is reduced because of the lower heat loss in the smaller nett flue gas flow. The case of the steam/natural gas reformer furnace is interesting as this only requires radiant heat and thus the higher emissivity furnace gas allows the firing rate to be reduced by 15% and still maintain the same radiant heat flux. The lower firing rate reduces the excess steam production in the convection section.
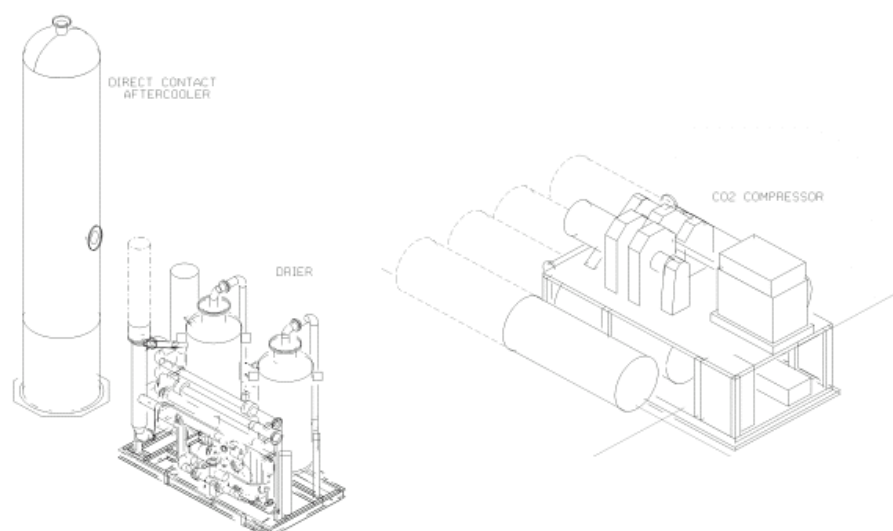
5.   Flue Gas Inerts Separation and CO_2 Compression

Due to the widely scattered location of the boilers and heaters in the refinery, it is necessary to collect the CO_2 rich flue gas and pipe it to a central location for final purification and compression.

The flue gas product from each oxyfuel unit varies in temperature from 180°C to 398°C and contains about 30% water vapour. The ambient pressure flue gas is piped a short distance to a local collection point where it is cooled in a venturi water scrubber followed by a direct contact packed tower. The water vapour is condensed and the flue gas is compressed in a centrifugal integrally geared compressor to 32 bar. The gas is then passed into a dual bed desiccant drier to reduce the water content to a dewpoint below -60°C. A PFD for the largest local CO_2 dryer, in the boiler area, is shown in Figure 2a alongside an isometric view of the equipment, Figure 2b. The dry CO_2 can now be piped in carbon steel lines to the central purification and final compression point.

Cooling water
2617 tonne/hr
24°C

Direct Contact Cooler

Boiler feed water
12.2 tonne/hr

Flue gas (collected from local heaters or boilers)

Compression
14.5 MW

Water knockout
2.4 tonne/hr

Dryers

Cooling water return
2704 tonne/hr
44°C

To central purification system

T, °C	283
kmol/hr	**8,777**
CO$_2$	40.77 mol%
O$_2$	1.67 mol%
Ar	2.10 mol%
N$_2$	8.08 mol%
H$_2$O	47.28 mol%
SO$_2$	0.08 mol%

Total, kmol/hr	**4,577**
T, °C	30.00
P, bara	32.06
CO$_2$	77.19 mol%
O$_2$	3.21 mol%
Ar	4.03 mol%
N$_2$	15.49 mol%
H$_2$O	0.00 mol%
SO$_2$	0.08 mol%

a) PFD of local CO$_2$ treatment area for boilers

DIRECT CONTACT AFTERCOOLER

DRIER

CO2 COMPRESSOR

b) Isometric view of area

Figure 2: PFD and Isometric view of CO$_2$ treatment area for boilers

Flue Gas Vent

Warm Exchanger

Cold Exchanger

PIC — AIC CO2

Flue Gas Heater

Flue Gas Expander

9.8 bara
-55.8 °C

CO$_2$	25.1 mol%
N$_2$	47.8 mol%
Ar	13.6 mol%
O$_2$	13.5 mol%
SO$_2$	0.0 mol%

Dry CO$_2$ from local drying and compression areas

19.6 bara
-30.9 °C

P	30 bara
CO$_2$	77.0 mol%
N$_2$	14.3 mol%
Ar	4.5 mol%
O$_2$	34.1 mol%
SO$_2$	0.1 mol%

CO$_2$ Compressor

AIC CO2

CO$_2$ Compressor

CO$_2$ product for sequestration

P	221 bara
CO$_2$	96.2 mol%
N$_2$	1.9 mol%
Ar	1.1 mol%
O$_2$	0.7 mol%
SO$_2$	0.1mol%

Figure 3: Central CO$_2$ purification system

Figure 3 shows an inerts gas removal plant using CO_2 refrigeration, which separates the inert gases from the CO_2 at a temperature of about $-55°C$ which is close to the CO_2 freezing temperature. At this point the CO_2 partial pressure in the vapour phase has been reduced to about 7 bar. The refrigeration is obtained by evaporating two streams of CO_2 at pressure levels of typically around 10 bar and 20 bar and recycling the CO_2 gas in the main CO_2 compressor. The separated inert gas at 29 bar can be heated and passed through a power recovery turbine. It is possible to reach a CO_2 purity in excess of 96% using this method at inlet CO_2 concentrations as low as 80% with a CO_2 recovery of better than 90%.
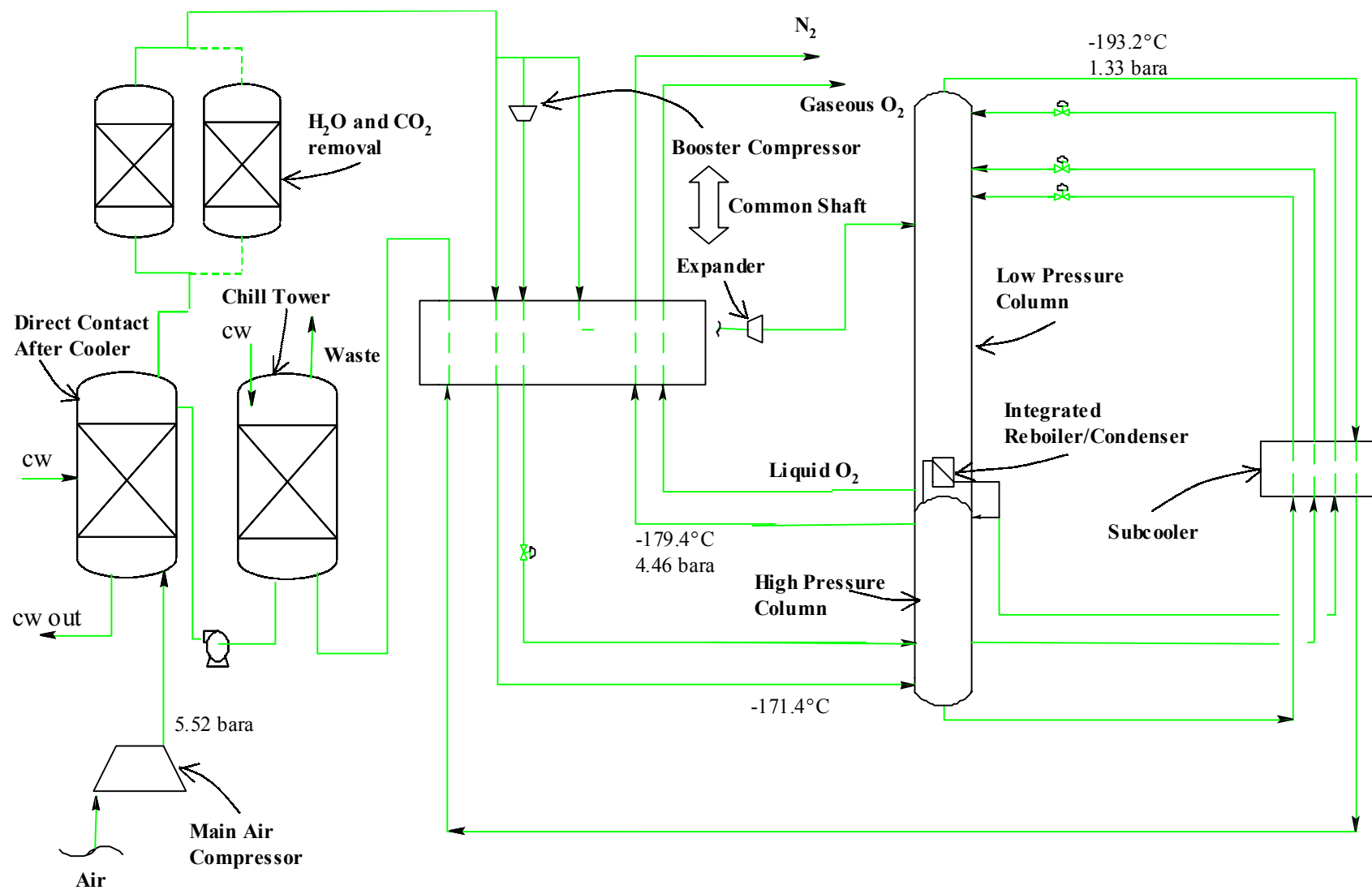
Figure 4: Cryogenic Air Separation Plant

6.   Cryogenic Air Separation Plant

The maximum total oxygen demand of 7400 tonne/day which includes a 10% flow margin is provided by two cryogenic air separation plants (ASUs) with single air compressors, Figure 4, provided as two trains of 3700 tonne/day, which is close to the current largest plant size of 3500 tonne/day.   The oxygen is delivered at 95% purity 1.7 bar abs into a pipeline system which runs to each of the oxyfuel use points. The plants utilise a cryogenic distillation system for air separation based on the use of an upper low pressure column in which the air is separated into a gaseous nitrogen stream leaving the top and a liquid oxygen stream leaving the base. The lower column is linked to the upper column through a reboiler-condenser in which N_2 separated from the air feed is condensed against boiling oxygen. The liquid nitrogen produced provides reflux for the upper and lower columns.

The reboiler conditions impose a pressure on the air feed of 5.5 bar abs.   The liquid oxygen is vaporised and the gaseous nitrogen and oxygen products are warmed to atmospheric temperature against cooling air, part of which has been further compressed to assist in the vaporisation of the oxygen. The heat exchange system uses aluminium plate-fin heat exchangers. The columns use a structured packing for separation of components.

Plant refrigeration is provided by an air expansion turbine discharging into the lower pressure column. Oxygen pressurisation is achieved using hydrostatic head, no liquid oxygen pumps or oxygen gas compressors are required. The air leaving the compressors is passed through a set of two switching adsorber vessels which remove water, CO_2, N_2O and hydrocarbon impurities which would contaminate the cryogenic equipment.

Each plant is provided with a single air compressor. These can be designed either as intercooled machines with an aftercooler, in which the heat of compression is rejected to cooling water or they can be specified as adiabatic compressors in which case they would be provided with an aftercooler which was used to heat boiler feed water from the condensers to 120°C before it was sent to the deaerators as shown in Figure 5. This will save 77 tonne/hr of 13.7 barg steam at present used for preheating condensate feed to the boilers. This steam can then be condensed to produce more power leading to a nett reduction in the overall power requirement for the air separation units.

This heating duty is possible for oxyfuel systems because the much reduced flue gas flowrate means there is a deficiency of low grate heat available. In each case the waste nitrogen from the cold box is used to chill water by evaporation in a direct contact tower and this chilled water is used for final air cooling before it enters the adsorbers. The overall air separation system performance is shown in Table 2 for the two cases. No cooling water is required for the adiabatic compressor case except at startup.
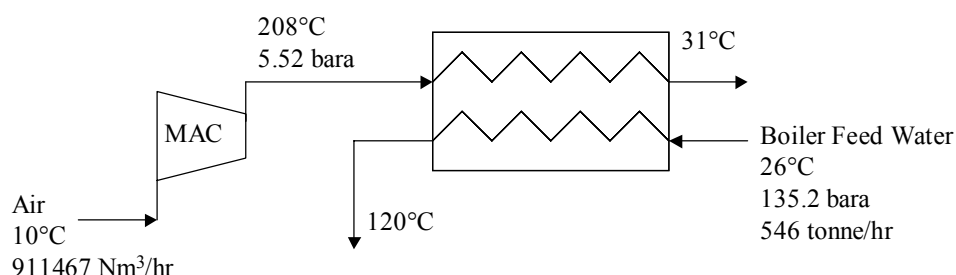

Figure 5: Adiabatic MAC (Main Air Compressor) integrated with condensate preheating for boiler feed

O2 flow tonne/day	Air Flow Nm3/hr	Air Pressure bara	Compressor Type	Power MW	Cooling Water Flow tonne/hr	Condensate Flow tonne/hr	13.7 barg Steam Turbine MW	Nett Power MW
6,736	911,467	5.52	Isothermal	63.8	7,819	0	0	63.8
6,736	911,467	5.52	Adiabatic	71.9	0	546	13.9	58.0

Table 2: ASU Performance Comparison

7. Utility Requirements

The oxyfuel conversion study includes the provision of all additional site services required for this area including cooling water and power production. Power will be provided using a gas turbine combined cycle system. Once a gas turbine model has been selected, the excess power can be fed into the refinery system. There are three options for dealing with the gas turbine CO_2 emissions and steam production.

1. The gas turbine and associated steam production is all used for power production.
2. The steam production is primarily at the 127 bar level and is used to replace part of the boiler steam, thus saving oxygen flow to the boilers.
3. The gas turbine could be run in the precombustion decarbonisation mode with part of the oxygen being used for hydrogen production in an autothermal reformer and with shift conversion and CO_2 removal using a MDEA system. For this case we have assumed excess steam production sent to the refinery turbines.

The overall performance of the complete system for these options is given in Table 3.

Case	GT	Power, MW GT	Power, MW ST	Fuel to Power System MW	Reduction in Fuel to Boilers MW	Reduction in Fuel to Heaters MW	CO_2 from GT exhaust x10^6 tonne/yr	Boiler Steam, x10^3 lb/hr From Boilers	Boiler Steam, x10^3 lb/hr From GT HRSG	Extra 13.7 barg Steam tonne/hr	Extra 13.7 barg Steam MW
1	6FA	70.1	37.0	202	27.75	70.30	0.374	1771	0	0	0
2	7EA	85.4	0	260	135.26	70.30	0.455	1483	287.6	24.2	4.4
3	7EA	97.0	4.5	302	100.64	70.30	0.022	1576	195.0	37.9	6.8

Case	Total Power Generated MW	O_2 required tonne/day	O_2 to boilers tonne/day	Total Power Requirement MW	Export Power MW	IGCC Cooling Water tonne/hr	Total Cooling Water tonne/hr	CO_2 for sequestering x10^6 tonne/yr	Reduction in CO_2 emitted x10^6 tonne/yr
1	107.1	6736	4316	95.3	11.8	6405	29,267	1.90	1.70
2	89.8	6035	3615	85.4	4.4	0	19,266	1.70	1.61
3	108.3	6890	3841	106.8	1.5	0	21,994	2.34	2.02

Table 3: Comparison of different power generation options

8.   Site Layout

Figure 6 shows the layout of the whole system in the refinery with the relative location of the oxyfuel systems, the air separation units and the cooling towers, together with the CO_2 and oxygen piping runs superimposed on a grid to show typical spacings and piping runs required.

9.   Conclusions

This work has shown is that it is feasible to apply Oxyfuel technology to a complete refinery system with multiple CO_2 emissions points spread out over a large area. This involves a centralised oxygen supply system and a CO_2 recovery, purification and compression facility.

It has been found that primary effluent gas cooling, compression and drying is best decentralised to be close to the emission points and an intermediate pressure CO_2 stream can then be routed to a centralised collection point for final purification and compression to pipeline pressure. The CO_2 purification system can be designed to handle practical levels of air infiltration into boilers and process heaters to produce a purity of CO_2 suitable for geological sequestration

The level of air infiltration into boilers and heaters that are retrofitted for Oxyfuel means that it is more economic to design the air separation units for only 95% purity and reject the associated argon and nitrogen in the CO_2 inert gas removal system.

It is possible to integrate the air separation system and the refinery steam system by using an adiabatic air compressor with boiler feed-water preheating in the compressor aftercooler. This minimises requirements for cooling water and also reduces overall power consumption.

All additional power consumption has been provided by a new gas turbine combined cycle system.

References

1. M. B. Wilkinson, JC Boden, T Gilmartin, C Ward, D. A. Cross, R. J. Allam and N.W.Ivens. "CO_2 Capture from Oil Refinery Process Heaters through Oxyfuel Combustion". (The 6th Greenhouse Gas Technologies Conference, October 2002, Kyoto, Japan.)
2. MB Wilkinson, JC Boden, R Panesar and R Allam.   "CO_2 Capture via Oxyfuel Firing: Optimisation of a Retrofit Design Concept for a Refinery Power Station Boiler." (First National Conference on Carbon Sequestration, May 15-17 2001, Washington DC.)
3. MB Wilkinson, JC Boden, R Panesar and R Allam.   "A study on the capture of carbon dioxide from a large refinery power station boiler by conversion to oxyfuel operation." (The 5th Greenhouse Gas Technologies Conference. August 2000, Cairns, Australia.)
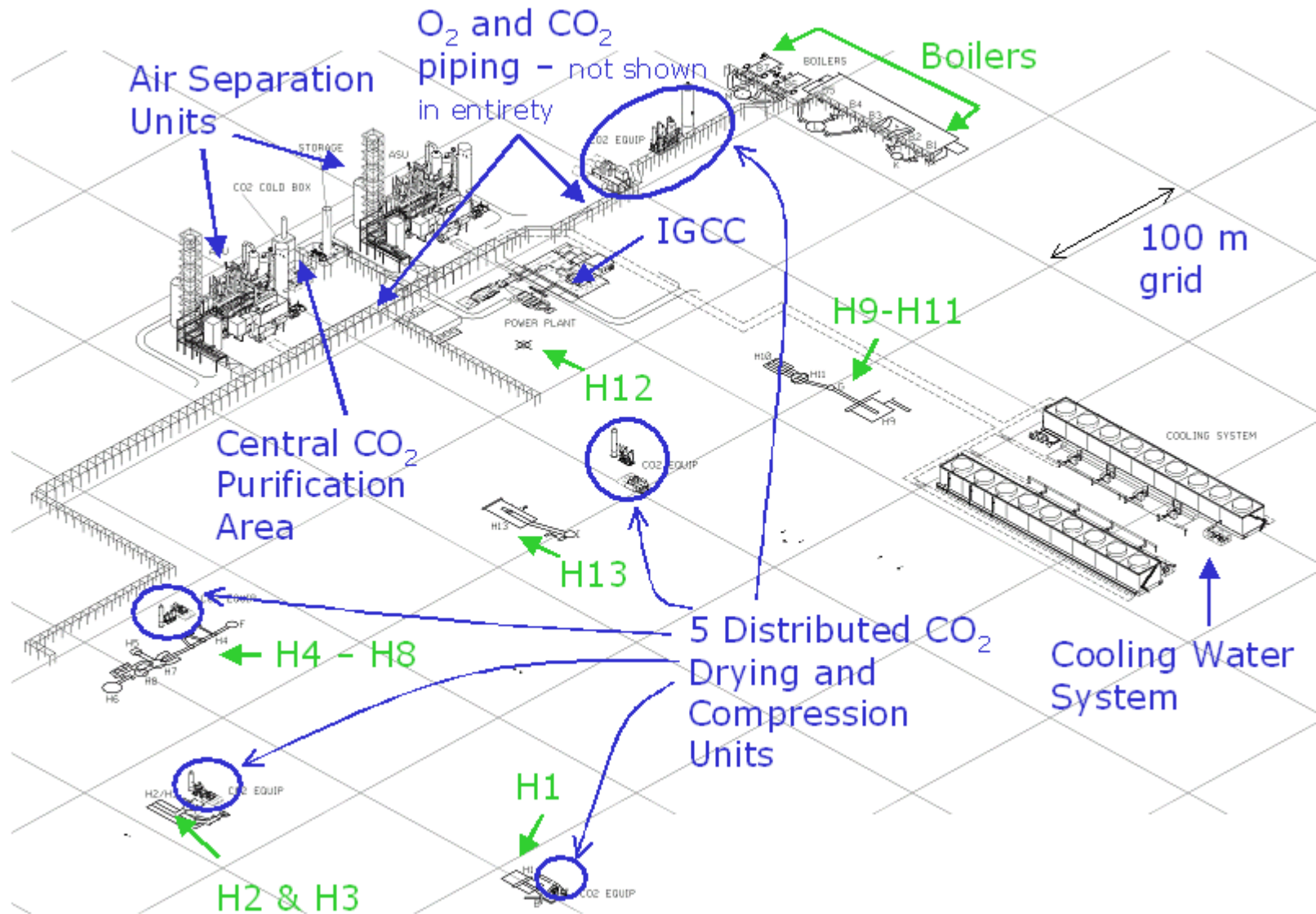
Figure 6: Isometric view of site layout showing the relative location of the oxyfuel systems, the air separation units and the cooling towers, together with the CO_2 and oxygen piping runs.